# STRUCTURES AND SYNTHESIS OF SEED-GERMINATION INHIBITORS FROM HIBISCUS ROSA-SINENSIS

MUNEHIRO NAKATANI, TAKASHI YAMACHIKA, TAKAO TANOUE\* and TSUNAO HASE

Department of Chemistry, Faculty of Science, Kagoshima University, Kagoshima 890, Japan; \*Oshima High School, Naze, Kagoshima 894, Japan

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**Abstract**—8-Nonynoic, 9-decynoic acids and their methyl esters were isolated for the first time as natural products from *Hibiscus rosa-sinensis*. They were inhibitors of the germination of lettuce seeds and were synthesized via o-nitrophenyl selenides.

## INTRODUCTION

The leaves and roots of hibiscus are widely used for medical purposes. In the course of an investigation of bioactive compounds of the plant, the stem bark of Japanese Hibiscus rosa-sinensis has been found to yield two acetylenic straight-chain acids, 8-nonynoic acid (3) and 9-decynoic acid (4), and their methyl esters, 1 and 2, which were inhibitors of the germination of lettuce seeds. They are one of the few classes of aliphatic compounds possessing a terminal acetylenic linkage from higher plants. The present paper describes the isolation, characterization and synthesis of these compounds.

# RESULTS AND DISCUSSION

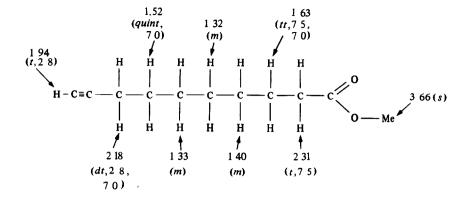
The isolation of 1–4, guided by bio-assay against lettuce seed, required careful use of HPLC. Compounds 1 and 2 were finally obtained by using 50% dichloromethane in hexane as the solvent: 1, oil,  $C_{10}H_{16}O_2$  (FDMS m/z: 168, [M]<sup>+</sup>); 2, oil,  $C_{11}H_{18}O_2$  (FDMS m/z: 182, [M]<sup>+</sup>). On the other hand, the use of 0.4–0.6% methanol in dichloromethane as the solvent gave a mixture of 3 and 4, which, on methylation with diazomethane, afforded 1 and 2 in a ratio of 4:1. Isolation of 3 and 4 was achieved using HPLC with a water-methanol solvent system: 3,  $C_9H_{14}O_2$  mp 22–23°; 4,  $C_{10}H_{16}O_2$ , mp 24.5–25.5°.

The IR absorptions of 3300 and 2100 cm<sup>-1</sup> in the spectra of 1 and 2 indicated the presence of a terminal acetylenic linkage, which was also supported by the EIMS fragments of m/z 54 [CH<sub>2</sub>=CH-CH=CH<sub>2</sub>]<sup>+</sup> and 68 [Me-(CH=CH)<sub>2</sub>-H]<sup>+</sup> [1]. Their <sup>1</sup>H NMR spectra showed acetylenic protons at  $\delta$ 1.94 (1H, t, J = 2.5 Hz and 1H, t, J = 2.8 Hz), coupling with H-7 and H-8, respectively, and carbomethoxyl groups at  $\delta$ 3.66. Homonuclear decoupling of the 360 MHz <sup>1</sup>H NMR spectra allowed us to assign all of the peaks in the spectra and derive structures 1 and 2 (Fig. 1). On hydrogenation with palladium-carbon, 1 and 2 gave methyl nonanoate and methyl decanoate, respectively.

Compounds 1-4 showed 50% inhibition to lettuce seed germination at 5, 10, 20 and 40  $\mu$ g/ml of water, respectively. These effects are comparable to those of a well-

known acetylenic growth inhibitor, cis-dehydromatricaric acid [2]. Aliphatic compounds have been reported to have inhibitory activity on lettuce seed germination [3]. Reynolds showed that inhibitory activity increased with carbon chain length and the main structure-activity correlation was with lipophilic properties [3, 4]. Increased lipophilicity usually leads to increased inhibitory activity; thus the methyl esters 1 and 2 showed increased activities. Recently, a variety of organic acids have also been reported to inhibit pollen germination in Camellia sinensis [5], and inhibition was greatest with the C<sub>10</sub>- and C<sub>11</sub>- acids. It is of interest that the C<sub>10</sub>-compounds 2 and 4 have lower activities than the C<sub>9</sub>-compounds 1 and 3.

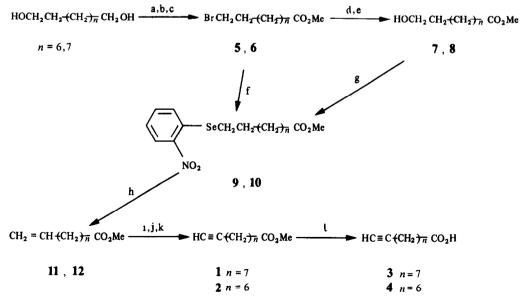
There are some reports on the synthesis of 9-decynoic acid (4) [6, 7], but we now deal with another approach to the synthesis of compounds 1-4 from 1,9- and 1,10-diols via o-nitrophenyl selenides 9 and 10 (Scheme 1). Recently, gentle formation of terminal olefins via the decomposition of primary alkyl phenyl selenides has been demonstrated by two groups [8, 9]. From 1,9-nonanediol and 1,10decanediol, 8- and 9-bromo-esters 5 and 6 were prepared in three steps in ca 35 % yields by mono-bromination with hydrobromic acid followed by oxidation and methylation. Treatment of 5 and 6 with o-nitrophenyl selenocyanate in the presence of sodium borohydride under nitrogen in ethanol at room temperature [8] gave o-nitrophenyl selenides 9, mp 61.5-62.5°, and 10, mp 44-45.5°, in 79 and 76% yields, respectively. On the other hand, alcohols 7 and 8 obtained by hydrolysis of 5 and 6 were also converted into the o-nitrophenyl selenides in 73 and 78 %yields, respectively, using o-nitrophenyl selenocyanate with tri-n-butylphosphine in tetrahydrofuran at room temperature [9]. Decomposition of these selenides by 30% hydrogen peroxide afforded olefins 11 and 12 in 85 and 86 % yields, respectively. The synthesis was completed by converting 11 and 12 into 8-nonynoic acid (3) and 9decynoic acid (4), by bromination with bromine and subsequent dehydrobromination with alcoholic potassium hydroxide under reflux, in 75 and 78% yields, respectively. Both methyl esters 11 and 12 inhibited the germination of lettuce seeds with only 50% germination at 1 ing/ml of water.



1 53 (quint, 1 42 2.31 (t,75)7 0) 1 94 (t, 25)Ĥ Н 3 66 (s) Ĥ H H H H 1 34 2 18 1 64 (dt, 25,(m) (quint, 70) 7.5) 2

1

Fig. 1. <sup>1</sup>H NMR data [CDCl<sub>3</sub>, 360 MHz, δ-values (ppm)] for 1 and 2, with multiplicities and J values (in Hz) given in parentheses.



Scheme 1. Synthesis of acetylinic acids. a, Aq. HBr; b, KMnO<sub>4</sub>, H<sub>2</sub>SO<sub>4</sub>; c, CH<sub>2</sub>N<sub>2</sub>; d, alcoholic KOH; e, CH<sub>2</sub>N<sub>2</sub>; f, o-nitrophenyl selenocyanate, (n-Bu)<sub>3</sub>P, THF; h, H<sub>2</sub>O<sub>2</sub>, THF; i, Br<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>; j, alcoholic KOH; k, CH<sub>2</sub>N<sub>2</sub>; l, alcoholic KOH.

#### **EXPERIMENTAL**

Mps are uncorr. UV spectra were recorded in MeOH solns.  $^1$ H NMR spectra of CDCl<sub>3</sub> solns with TMS as internal standard ( $\delta = 0$  ppm) were measured at 60 and 360 MHz.

Plant material. Stem bark of H. rosa-sinensis L. was collected in August 1980 at Amami-Oshima, Kagoshima and identified by Mr. M. Fukuyama (Kagoshima Agricultural Experiment Station).

Extraction and isolation. Fresh stem bark (2.6 kg) was extracted with MeOH  $(3 \times 101)$  and, after concn to 1.5 l., the MeOH extract was extracted with hexane to yield 24 g of an extract. This extract was chromatographed on silica gel with MeOH-CH<sub>2</sub>Cl<sub>2</sub> and the active fractions were rechromatographed on silica gel with Et2O-hexane to give two active fractions. Repeated passage of a non-polar fraction through an HPLC, Whatman Partisil M9 semiprep. column, using 50-70% CH<sub>2</sub>Cl<sub>2</sub> in hexane as solvent finally gave the following compounds: Me 8-nonynoate (1), 46 mg: oil,  $C_{10}H_{16}O_2$ ; FDMS m/z: 168 [M]<sup>+</sup>; EIMS m/z(rel. int.): 137 (5), 94 (53), 81 (23), 74 (100), 68 (13), 54 (22); IR  $v_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3300, 2100, 1725; UV  $\lambda_{\text{max}}$  nm ( $\epsilon$ ): 206 (140). Me 9-decynoate (2), 12 mg: oil,  $C_{11}H_{18}O_2$ ; FDMS m/z: 182 [M]<sup>+</sup>; EIMS m/z (rel. int.): 151 (8), 108 (66), 87 (64), 81 (94), 74 (100), 68 (56), 54 (19); IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>-1</sup>: 3300, 2120, 1725; UV  $\lambda_{\text{max}}$  nm ( $\epsilon$ ): 206 (140). By use of 0.4-0.6% MeOH in CH<sub>2</sub>Cl<sub>2</sub> as solvent, the polar fraction gave a mixture (15 mg) of 3 and 4, which was isolated on a Waters  $\mu$ -Bondapack  $C_{18}$  column with 53 %  $H_2O$  in MeOH: 8-nonynoic acid (3): mp 22-23°; IR  $v_{\text{max}}^{\text{liquid film}}$  cm<sup>-1</sup>: 3300, 3200-2500, 2100, 1703. 9-Decynoic acid (4): mp 24.5-25.5°; IR  $v_{\text{max}}^{\text{liquid film}}$  cm<sup>-1</sup>: 3300, 3200–2500, 2100, 1700.

Hydrogenation of 1 and 2. Compounds 1 (5 mg) and 2 (5 mg) were treated with Pd-C under H<sub>2</sub> gas in hexane to give Me nonanoate and Me decanoate, respectively. These compounds were identified by direct comparison with authentic samples (IR and <sup>1</sup>H NMR).

Methylation of a mixture of 3 and 4. A mixture (10 mg) of 3 and 4 obtained from CC was methylated with  $CH_2N_2$ - $Et_2O$  to give 1 (8 mg) and 2 (2 mg).

Hydrolysis of 1 and 2. In EtOH, 1 (5 mg) and 2 (5 mg) were gently refluxed with a small amount of 30% KOH soln for 30 min to give quantitatively the corresponding acids 3 and 4, respectively.

Synthesis of 9- and 10-bromo-esters 5 and 6. 1,9-Nonanediol (5 g) was treated in a soln of 48 % HBr (6.5 g) and conc.  $H_2SO_4$  (1.1 ml) for 3 hr under reflux.  $Et_2O$  extraction gave a crude product, which was oxidized with KMnO<sub>4</sub> (2.5 g) in 20 %  $H_2SO_4$  (28 ml) at 15-20°. After methylation of the crude product with  $CH_2N_2$ , the product was chromatographed over silica gel to give 5 (29 g): oil;  $IR \ v_{hqui}^{hquid film} \ cm^{-1}$ : 1740. <sup>1</sup>H NMR (60 MHz):  $\delta$ 2.27 (2H, t, J = 7 Hz,  $-CH_2$ -CO-), 3.38 (2H, t, J = 6.5 Hz,  $-CH_2$ Br), 3.65 (3H, s,  $-CO_2$ Me). In a similar manner as above, 1,10-decanediol (5 g) afforded 6 (2.8 g): oil;  $IR \ v_{hqx}^{hquid film} \ cm^{-1}$ : 1740. <sup>1</sup>H NMR (60 MHz):  $\delta$ 2.28 (2H, t, J = 7 Hz,  $-CH_2$ CO-), 3.38 (2H, t, J = 6.7 Hz,  $-CH_2$ Br), 3.66 (3H, s,  $-CO_2$ Me).

Methyl 9-hydroxynonanoate (7) and methyl 10-hydroxydecanoate (8). The bromo-esters 5 and 6 were hydrolysed with aq. 20% NaOH at 50° and the products were methylated with CH<sub>2</sub>N<sub>2</sub> to give 7 and 8 in ca 90% yields, respectively. Compound 7: mp 56.5-57.5°; IR  $\nu_{\text{max}}^{\text{liquid}}$  6im cm<sup>-1</sup>: 3400, 1740; <sup>1</sup>H NMR (60 MHz): δ2.27 (2H, t, J = 7 Hz, -CH<sub>2</sub>CO-), 3.33 (2H, t, J = 6 Hz, -CH<sub>2</sub>OH), 3.60 (3H, s, -CO<sub>2</sub>Me). (Found: C, 63.56; H, 11.06. Calc. for C<sub>10</sub>H<sub>20</sub>O<sub>3</sub>: C, 63.83; H, 10.72%) Compound 8: mp 59.5-60.5°; IR  $\nu_{\text{max}}^{\text{liquid}}$  6im cm<sup>-1</sup>: 3350, 1740; <sup>1</sup>H NMR (60 MHz): δ2.31 (2H, t, J = 7 Hz, -CH<sub>2</sub>CO-), 3.32 (2H, t, J = 6 Hz, -CH<sub>2</sub>OH), 3.67 (3H, s, -CO<sub>2</sub>Me); EIMS (rel. int.): 203 [M+1]<sup>+</sup> (19), 185 (12), 153 (25), 97 (44), 83 (100), 74 (13), 69 (92).

(Found: C, 65.28; H, 10.88. Calc. for C<sub>11</sub>H<sub>22</sub>O<sub>3</sub>: C, 65.35; H, 10.98 %.)

Preparation of selenides 9 and 10. (i) a. To a cooled (0°) suspension of o-nitrophenyl selenocyanate (1.03 g, 4.5 mmol) in EtOH (20 ml) under N<sub>2</sub>, NaBH<sub>4</sub> (0.19 g, 5 mmol) was added with stirring. Me 9-bromononanoate (5) (1.1 g, 4.4 mmol) was added and the soln stirred at room temp. for 13 hr. After removal of EtOH, the crude product was chromatographed over silica gel to give 1.3 g (79%) of selenide 9 as a yellow crystalline compound: mp 61.5-62.5°; IR  $v_{\text{max}}^{\text{Nujol}}$  cm<sup>-1</sup>: 1727, 1587, 1560. <sup>1</sup>H NMR (60 MHz):  $\delta 2.28$  (2H, t, J = 7 Hz,  $-\text{CH}_2\text{CO}$ -), 2.90 (2H, t, J= 7 Hz,  $-C\underline{H}_2\text{Ph}$ ),  $3.65 (3\text{H}, s, -CO_2\text{Me})$ , 7.27-7.69 (3H, m, Ph-<u>H</u>), 8.38 (1H, d, J = 8 Hz, Ph-<u>H</u>). (Found: C, 51.65, H, 6.23; N, 3.75. Calc. for C<sub>16</sub>H<sub>23</sub>NO<sub>4</sub>Se: C, 51.61; H, 6.23; N, 3.76%) b. In a similar manner as above, Me 10-bromodecanoate (6) (1.45 g, 7.3 mmol) afforded 1.55 g (76%) of 10 as a yellow crystalline compound: mp 44-45.5°; IR  $v_{\text{max}}^{\text{liquid film}}$  cm<sup>-1</sup>: 1733, 1590, 1560. <sup>1</sup>H NMR (60 MHz):  $\delta$ 2.32 (2H, t, J = 7 Hz,  $-\text{CH}_2\text{CO}-$ ), 2.96  $(2H, t, J = 7.5 \text{ Hz}, -CH_2\text{Ph}), 3.69 (3H, s, -CO_2\text{Me}), 7.36-7.72$ (3H, m, Ph- $\underline{H}$ ), 8.35 (1H, d, J = 8 Hz, Ph- $\underline{H}$ ). (Found: C, 52.77; H, 6.61; N, 3.65 Calc. for C<sub>17</sub>H<sub>25</sub>NO<sub>4</sub>Se: C, 52 89; H, 6.53;

(ii) a. A soln of Me 9-hydroxynonanoate (7) (450 mg, 2.4 mmol) in 8 ml of THF containing o-nitrophenyl selenocyanate (672 mg, 2.9 mmol) under  $N_2$  was treated dropwise with tri-n-butylphosphine (600 mg, 2.9 mmol) at room temp. After the reaction was stirred for 30 min, the solvent was removed in vacuo. Chromatography of the residue on silica gel gave 753 mg (73%) of 9. b. In a similar manner as above, 8 (725 mg, 3.65 mmol) afforded 1.1 g (78%) of 10.

Preparation of methyl 8-nonenoate (11) and methyl 9-decenoate (12). (i) Into a cooled (0°) soln of 9 (582 mg, 1.56 mmol) in THF (15 ml), 30 % H<sub>2</sub>O<sub>2</sub> (4 ml, 60 mmol) was added dropwise The ice bath was removed and the soln stirred for an additional 15 hr. The mixture was diluted with H<sub>2</sub>O and extracted with hexane. Chromatography of the hexane extract gave 225 mg (85%) of 11 as a very smelling oil: IR  $v_{\text{max}}^{\text{liquid film}}$  cm<sup>-1</sup>: 3070, 1740, 1640. <sup>1</sup>H NMR (60 MHz):  $\delta$ 2.27 (2H, t, J = 6.5 Hz,  $-\text{CH}_2\text{CO}-$ ), 3.66 (3H, s, -CO<sub>2</sub>Me), 4.78 (1H, brd, J = 8 Hz,  $^{H}$ >C=C $< ^{H}$ ), 4.83 (1H, brd, J = 16 Hz,  ${}^{H}>C=C<_{H}^{H}$ ), 5.40-6 10 (1H,  ${}^{H}>C=C<_{H}^{H}$ ). FDMS m/z: 170 [M]<sup>+</sup>. (ii) In a similar manner as above, 10 (1.1 g) afforded 12 (440 mg, 84%) as a smelling oil; IR  $v_{\text{max}}^{\text{liquid}}$ film cm<sup>-1</sup>: 3060, 1740, 1638. <sup>1</sup>H NMR (60 MHz):  $\delta$ 2.27 (2H, t, J = 6.5 Hz,  $-\text{CH}_2\text{CO}$ -),  $3.63 \text{ (3H, s, } -\text{CO}_2\text{Me)}$ , 4.78 (1H, br d, J)= 9 Hz,  $^{\text{H}}>\text{C}=\text{C}<\frac{\text{H}}{\text{H}}$ ), 4.83 (1H, brd, J=16 Hz,  $^{\text{H}}>\text{C}=\text{C}<\frac{\text{H}}{\text{H}}$ ), 5 40-6.11(1H,  $\stackrel{\text{H}}{>}$ C=C $\stackrel{\text{H}}{<}$ I. FDMS m/z: 184 [M] $^+$ .

Preparation of 1 and 2 from 11 and 12. To a soln of 11 (480 mg) in CHCl<sub>3</sub> (10 ml), excess Br<sub>2</sub> was added and the soln stirred at room temp. for 1 hr. After usual work-up, the product was treated with KOH (670 mg) in EtOH under reflux for 1.5 hr. After acidification, the soln was extracted with hexane and the extract methylated with CH<sub>2</sub>N<sub>2</sub> to give 1 (356 mg, 75 %), which showed the same spectral properties (IR and  $^{1}$ H NMR) and biological activity as the natural product. In a similar manner as above, 12 afforded 2 (78 %).

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